

# Durcissement des aciers inoxydables ferritiques : Thermi-SNT

**47<sup>ème</sup> Congrès A3TS - Marseille 2021**

## High Temperature Nitriding of Ferritic Stainless Steels : Thermi-SNT

# → THERMI-LYON GROUP

**160**  
EMPLOYEES

**+ 5 000**  
CUSTOMERS

**100 %**  
QUALITY-CERTIFIED SITES

**+20 M€**  
CONSACRATED  
INNOVATION AND  
INVESTMENT  
OVER THE LAST 4 YEARS

**23 M€**  
IN TURNOVER

THERMI-LYON has been able to adapt to changes in its markets by diversifying its scope. Today, the group is a reference in heat treatment and vacuum coatings.

- 8 national sites
- 1 international site

# → Outline

- Context and Background
- Key parameters and inputs
- Simulation
- Characterisations
- Conclusion



# → Context

- Market expectation :
  - Stainless Steel + Abrasive and Adhesive Wear Resistance
- Solutions :
  - Use **martensitic stainless steel** in the hardened state – But corrosion resistance is sometimes not sufficient for some applications
  - Use **austenitic or ferritic stainless steel** in work-hardened state – But wear and adhesive wear resistance is not sufficient for most of applications.
  - Perform a **surface hardening** by a **thermochemical treatment** under conditions preserving the corrosion resistance :
    - For austenitic steels : Low T° Nitrocarburizing = **Thermi®-SP**
    - For martensitic or ferritic steels : High T° Nitriding = **Thermi®-SNT**

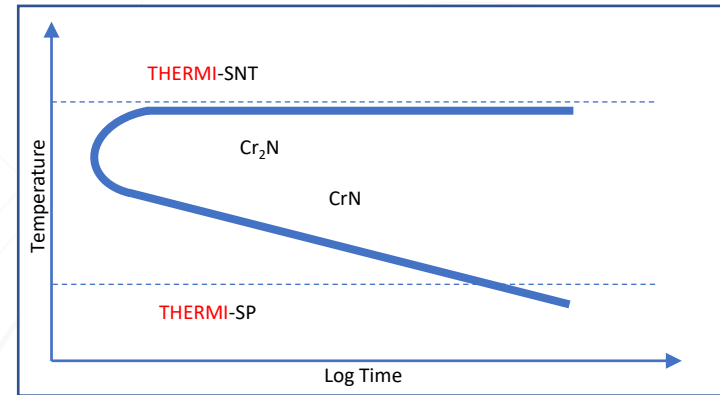
# → Background

- The first works on the "Solution Nitriding" process was published in 1994 by Hans Berns (Ruhr-Universität, Bochum, D).
- Commercial communications by the furnace supplier Ipsen on this treatment (branded Solnit<sup>©</sup>) have been found for about 15 years.
- Thermi-Lyon is equipped with IPSEN furnaces incorporating this technology. For the consistency of its treatment range, Thermi-Lyon has named this treatment **Thermi<sup>©</sup>-SNT**
- The purpose of this talk today is to describe the approach used to control this process available on the market and the results obtained.



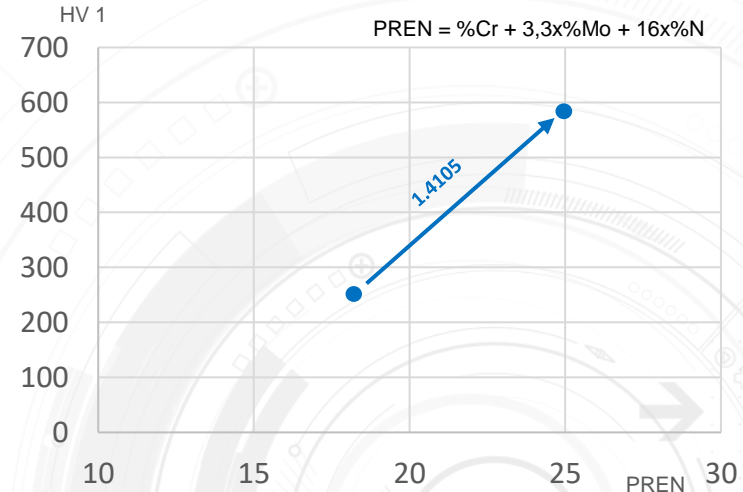
# → Key Parameters

- The corrosion resistance of stainless steels is based on their ability to form a passive layer based on oxidized chromium.
- The surface enrichment in N requires activating the steel surface by first reducing the passive layer.
- The treatment conditions must then be chosen so as to avoid the precipitation of chromium nitrides which, if they form, consume the free chromium responsible for the passivation.
- Once enriched, controlling the cooling conditions and the re-passivation of stainless steel guarantee good corrosion resistance.



# → Inputs to treat Ferritic Stainless Steels

- **Thermi**©-SNT Process T° range is 1050-1200°C
- T° is high enough to dissociate molecular N<sub>2</sub> into atomic nitrogen N
- N diffusion brings about a local ferrite **transformation in austenite**.
- During the fast cooling, Austenite is transformed to **Nitrogen Martensite** that leads to surface hardening
- Topic today is to adjust treatments parameters to the **steel grade 1.4501 (AISI 430F - X6CrMoS17)**



# → Thermo-Calc Simulation



*All calculations were carried out by the LMI Labs at Claude Bernard University, Lyon*

- **CalPhaD Méthod** (Calculation Phase Diagram)
  - Free Gibbs Energy (G) calculation f ( $T^\circ$ , P, steel Composition).
  - Adjusting G to a minimum value to identify the more stable states and to construct the equilibrium diagram
- Databases used are TCFE8 and TCFE10
- Nitrogen activity is calculated for several  $N_2$  partial pressures
- Thermodynamic equilibrium between the gas and the solid is assumed to be obtained
- Diagrams plotted are : %N mass = f(activité) and %N mass = f( $P_{N_2}$ )



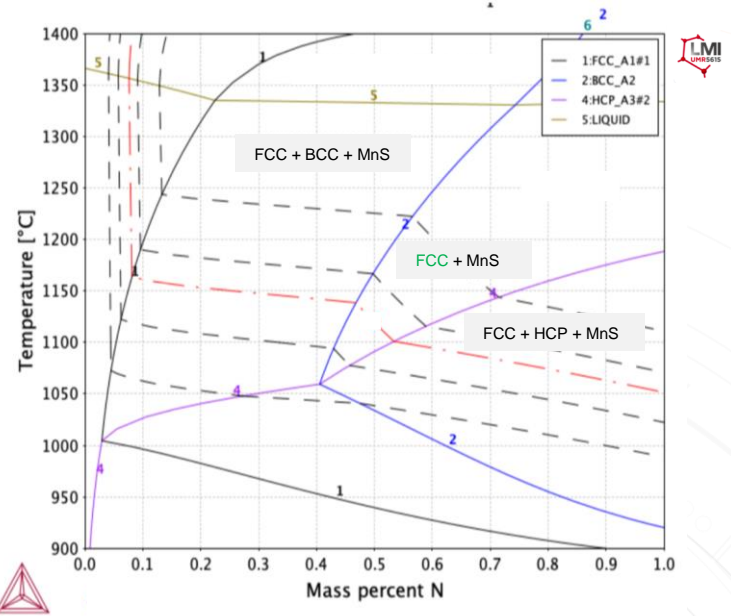
# → Composition and Equilibrium Diagram

## Steel batch n°1

1.4105 / AISI 430F	%C	%Si	%Mn	%P	%S	%Cr	%Ni	%Mo	%Fe
X6CrMoS17 NF EN 10088-3	0.08 max	1.50 max	1.50 max	0.040 max	0.15-0.35	16.0-18.0	-	0.20-0.60	Bal.
X6CrMoS17 COMPO 1 étudiée	0.005	1.214	0.505	0.020	0.278	17.44	-	0.273	Bal.

1.4105 Chemical composition Batch n°1 (mass%)

Both T° and N<sub>2</sub> Pressure must be chosen to be correctly positioned in the diagram in order to obtain 100% FCC and to avoid the precipitation of chromium nitride.



Vertical section (T, %m N, pN<sub>2</sub>) 1.4105 – Batch n°1

FCC = austenite / HCP = CrN / BCC = ferrite / MnS = Manganese Sulphide



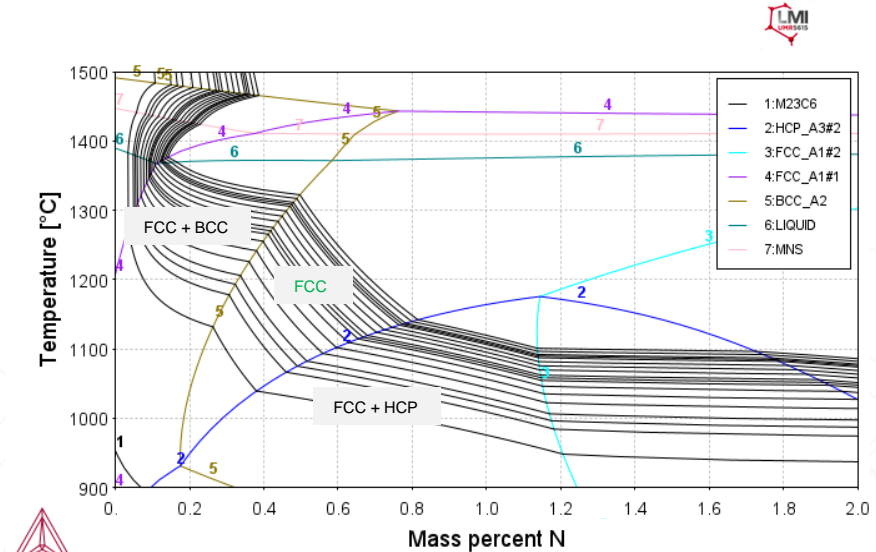
# → Composition and Equilibrium Diagram

## Steel batch n°2

1.4105 / AISI 430F	%C	%Si	%Mn	%P	%S	%Cr	%Ni	%Mo	%Fe
X6CrMoS17 NF EN 10088-3	0.08 max	1.50 max	1.50 max	0.040 max	0.15-0.35	16.0-18.0	-	0.20-0.60	Bal.
X6CrMoS17 COMPO 1 étudiée	0.005	1.214	0.505	0.020	0.278	17.44	-	0.273	Bal.
X6CrMoS17 COMPO 2 étudiée	0.064	0,523	0,684	0,020	0,280	16,17	0,183	0,226	Bal.

1.4105 Chemical composition Batch n°2 (mass%)

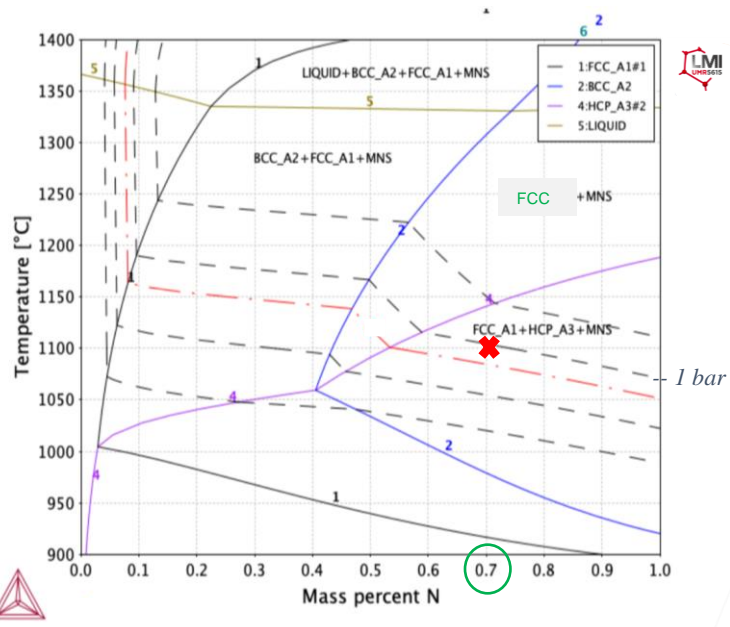
The difference in composition leads in a significant shift in the boundaries between domains.



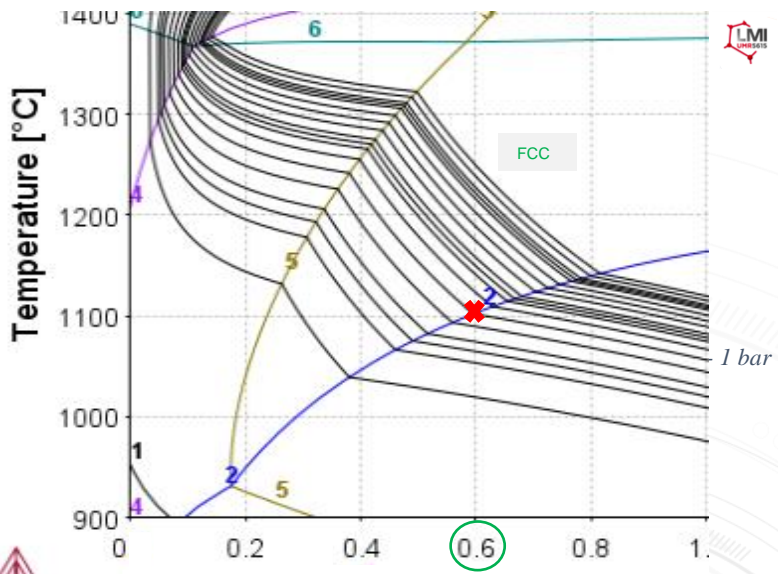
Vertical section (T, %m N, pN2) 1.4105 – Batch n°2

FCC = austenite / HCP = CrN / BCC = ferrite

# → Diagram comparison



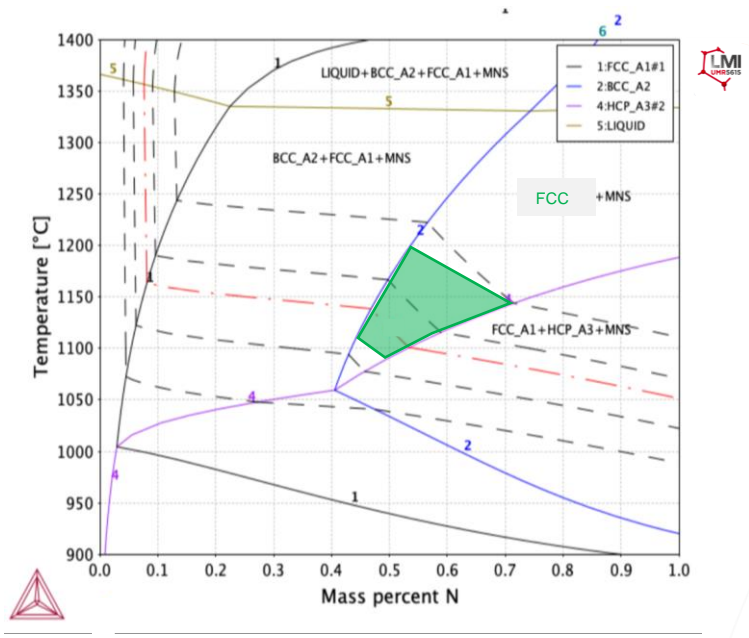
Steel Batch n°1



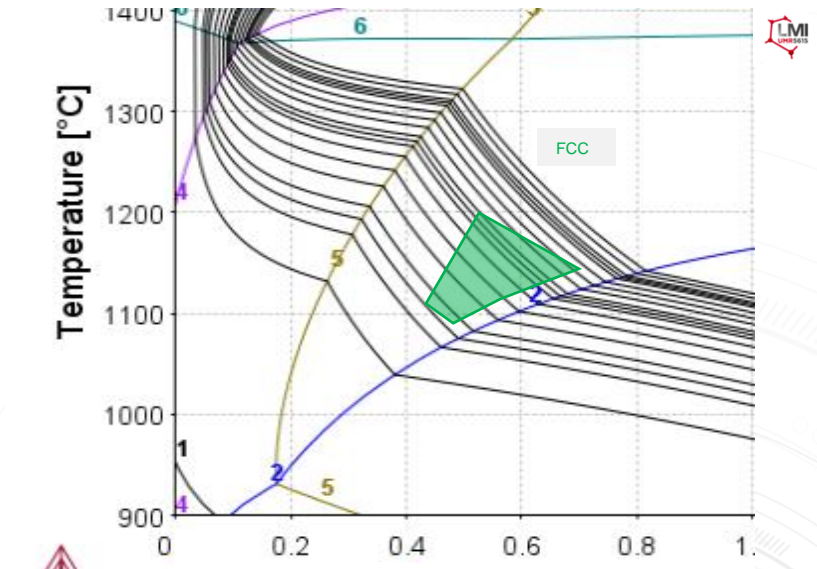
Steel Batch n°2

FCC = austenite / HCP = CrN / BCC = ferrite / MnS = Manganese Sulphide

# → Thermi<sup>©</sup>-SNT parameters compatible with both comp.



Steel Batch n°1



Steel Batch n°2

FCC = austenite / HCP = CrN / BCC = ferrite / MnS = Manganese Sulphide

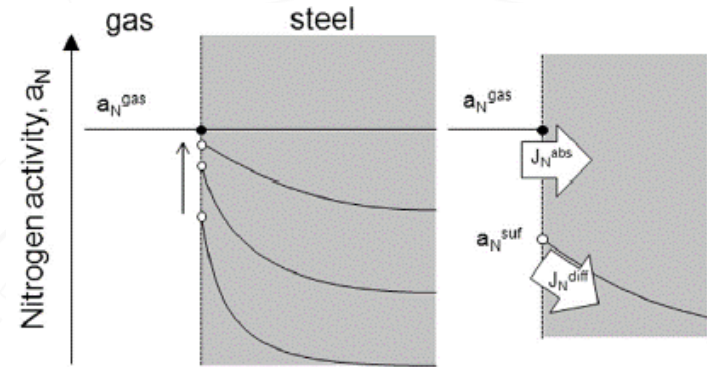
# → Prediction of Diffusion Profiles vs Time

- N diffusion is driven by Fick's Laws
- Activity as a flux function is a more accurate approach than a fixed activity. (Ref. Nakada, Kyushu University, Japan - 2014)

$$\rightarrow J_N^{abs} = k\rho(a_N^{gaz} - a_N^{sur}) = f(a_N^{gaz} - a_N^{sur})$$

$$\text{Avec } f = k\rho = 1.10 \cdot 10^{-8} \text{ mol/m}^2 \text{ s}$$

$K$  = mass transfert coefficient (m/s) and  $\rho$  = the density of the steel (mol/m<sup>3</sup>)

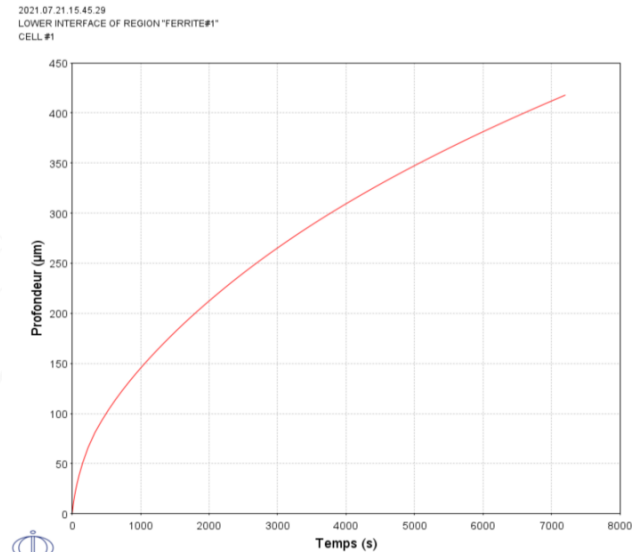
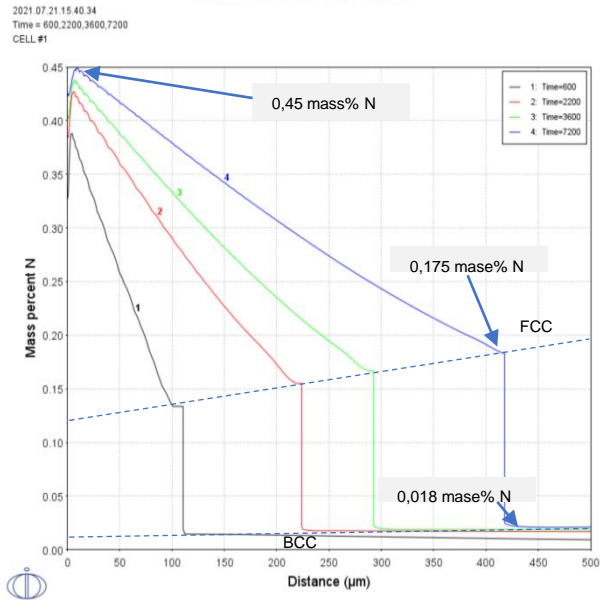


To predic the diffusion depth as a function of **Thermi**©-SNT conditions, calculation is performed using DICTRA software and the Database MODFE2

# → Dictra Predictions

**Thermi<sup>©</sup>-SNT** parameters ( $T^\circ$  and  $N_2$  pressure) optimised for 1.4105

*The vertical line = interface between FCC and BCC phases*



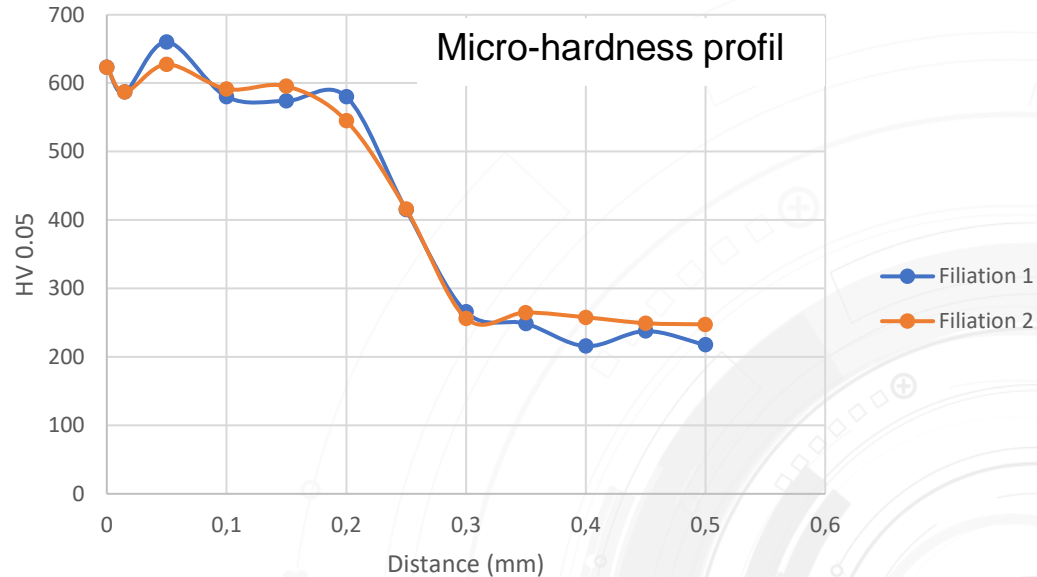


# → Characterisation after **Thermi**©-SNT

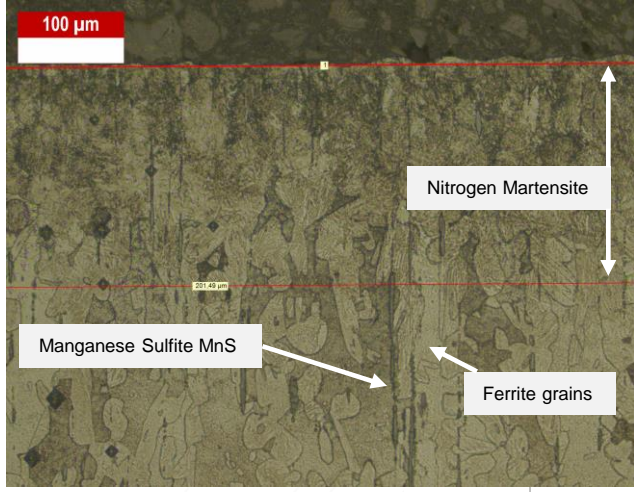
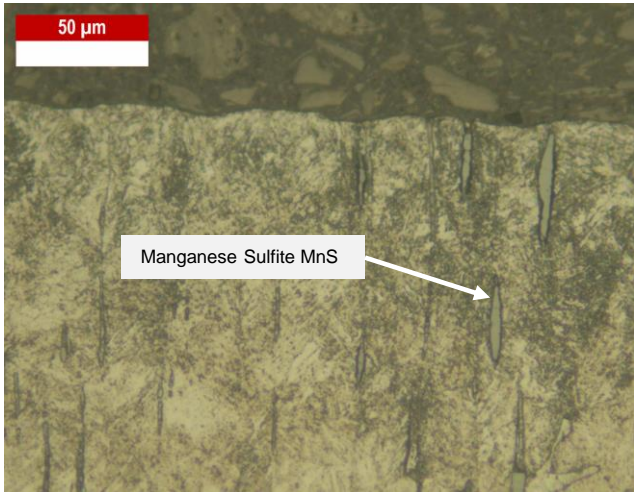
- Samples were characterised by
  - Hardness measurement (Superficial and Core)
  - Microhardness profiles
  - Optical Microscopy observations
  - X-Ray Diffraction
  
- Properties were evaluated by
  - Cyclic Voltametry for the corrosion behavior
  - ASTM G196 for the adhesive wear

# → Hardness and Thickness

HV0,05 sup	HV1 sup	HV10 cœur	Profondeur de filiation ( $\mu\text{m}$ )	Profondeur théorique par Dictra ( $\mu\text{m}$ )
623	588	261	277	226

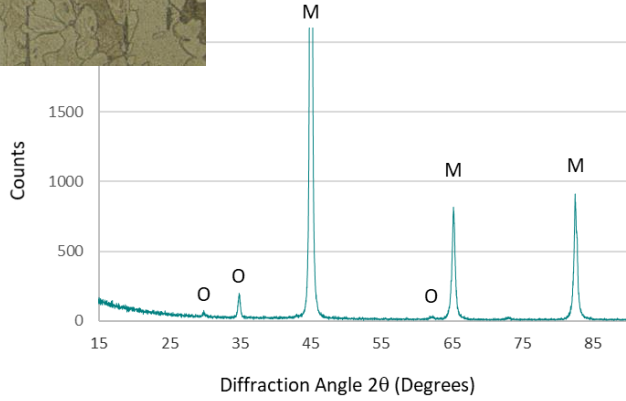


# → Typical Surface Microstructure



Etching : Aqua Regia.

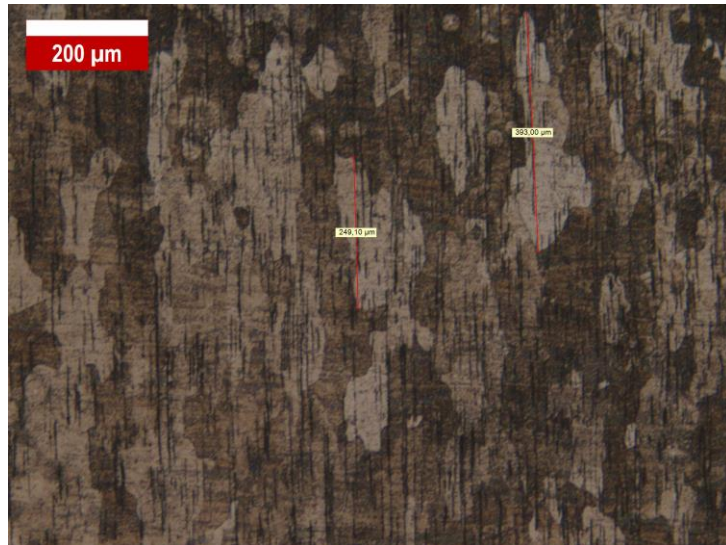
No Chromium Nitrides observed



# → Core Microstructure

Comparison of core microstructures before and after treatment :

→no grain growth observed

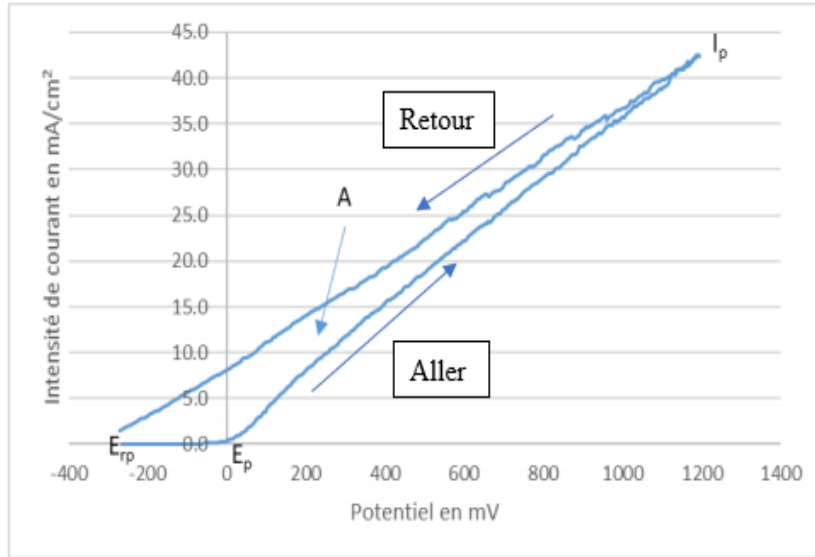


*Core microstructure before treatment*

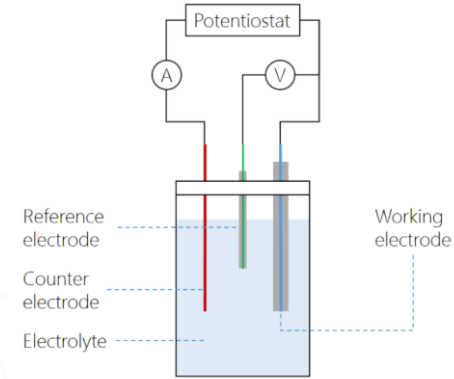


*Core microstructure after treatment*

# → Cyclic Voltammetry



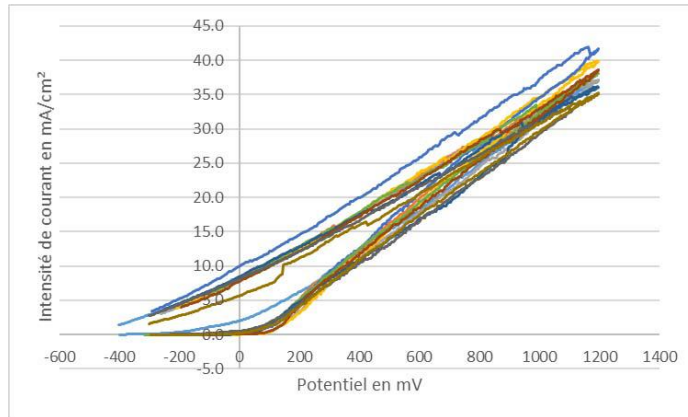
Exemple of curves obtained by cyclic voltammetry



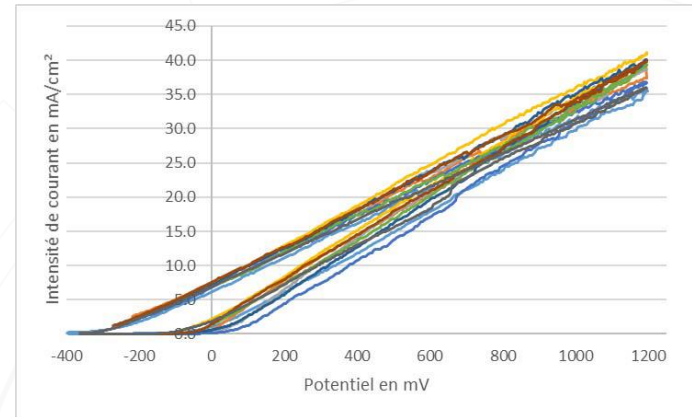
- *Potential  $E_p$  : Start of pitting*
- *Intensity  $I_p$  : proportional to the number of pits or their severity*
- *Potential  $E_{rp}$  : when pits are neutralized*
- *Area A : the weaker A, the stronger pitting resistance*
- *Slope  $E_p - I_p$  represents the corrosion kinetics*

# → Cyclic Voltammetry

- **Thermi<sup>©</sup>**-SNT samples are slightly more resistant to pitting than untreated ones
- Pitting Potential  $E_p$  is similar but with half less extensive values
- The corrosion kinetics is slightly lower after treatment



Before treatment



After treatment

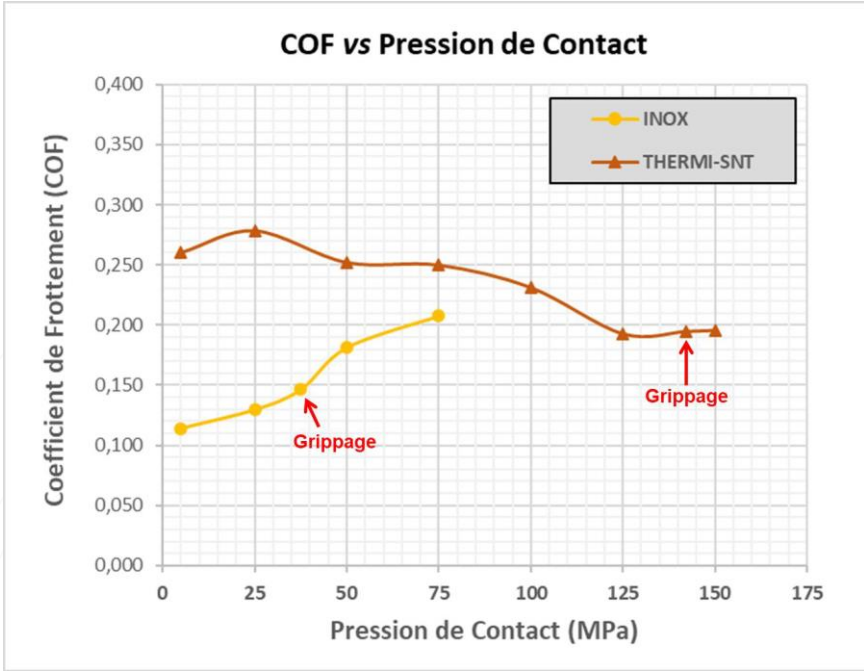


# ➔ Adhesive Wear Evaluation

- ASTM G196 Conditions

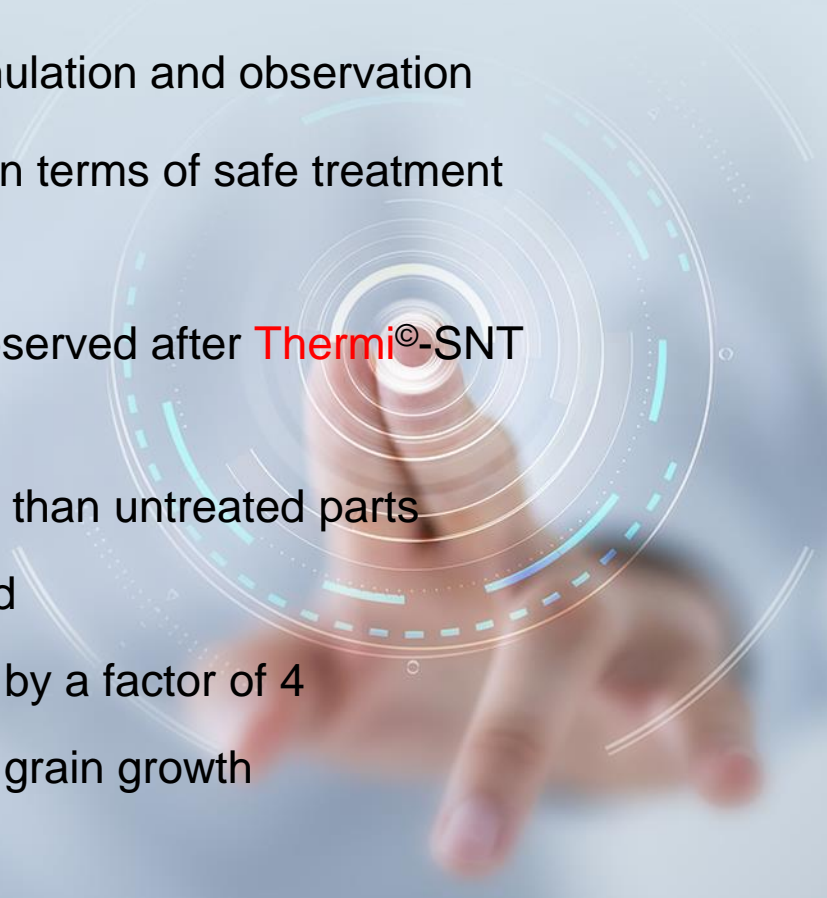
Valeur / Description	
Contact	Plan / Plan (Pion /Disque) Pion en INOX non revêtu Disque à tester en INOX, sans et avec Traitement de surface
Mouvement	Rotatif continu (1 Tour)
Pression de contact	De 5 MPa à 400 MPa
Vitesse de rotation	6 RPM
Durée de l'essai	10 s
Température	Ambiante (23 °C)
Lubrification	Contact sec

Palier de Grippage				
	Essai – 1	Essai – 2	Essai – 3	Moyenne
Inox non traité	25 MPa	50 MPa	-	37,5 MPa
Thermi-SNT	150 MPa	125 MPa	150 MPa	142 MPa



# → Conclusions

- There is a good consistency between simulation and observation
- Simulation saves time and is guaranteed in terms of safe treatment parameters
- Improvement of the steel grade 1.4105 observed after **Thermi**©-SNT treatment :
  - Superficial hardness is 2,5 time higher than untreated parts
  - Corrosion resistance is slightly improved
  - Adhesive wear resistance is improved by a factor of 4
  - Core microstructure is preserved from grain growth





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DES MATÉRIAUX



**THERMI-LYON**   
Groupe

*Merci pour votre attention*

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